

# Soil Fertility Management in Tropical Agriculture

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# Importance of smallholder farming

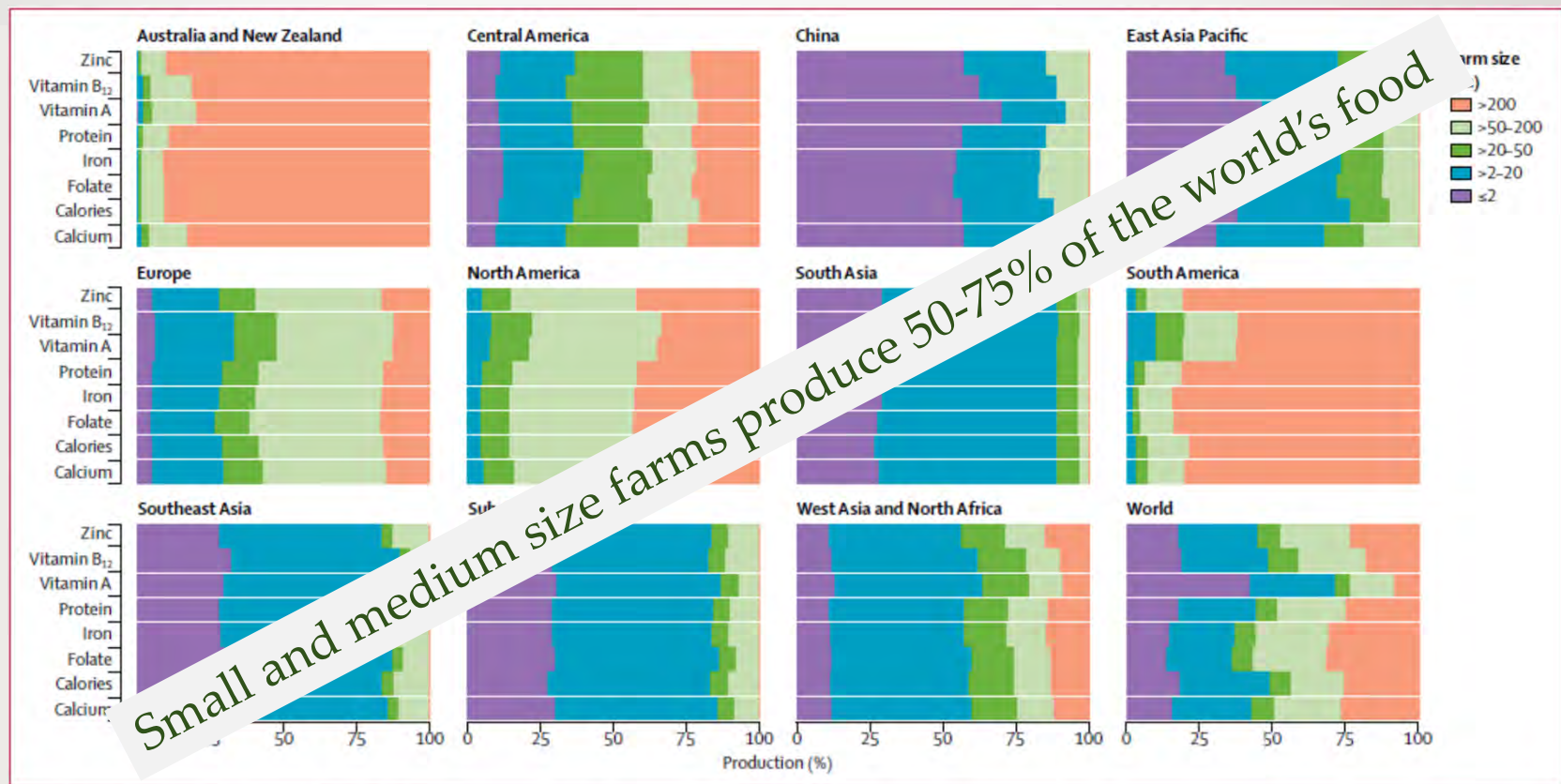


Figure 3: Distribution of nutrient production by farm size

Herrero et al., Lancet Planet Health 2017; 1: e33–42



Malawi



Kenya



China



Senegal

Haiti





SOIL



IS NOT

DIRT!

# Topics

- Essential nutrients for plant growth
- Formation and fertility of tropical soils
- Nutrient balance for crop production
- Improving soil fertility in smallholder farming

# Periodic Table of the Elements

	1 IA 1A												13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	18 VIIIA 8A
1	1 H Hydrogen 1.008 [He]1s <sup>1</sup>	2	3 Li Lithium 6.941 [He]2s <sup>1</sup>	4	4 Be Beryllium 9.012 [He]2s <sup>2</sup>								5 B Boron 10.811 [He]2s <sup>2</sup> 2p <sup>1</sup>	6 C Carbon 12.011 [He]2s <sup>2</sup> 2p <sup>2</sup>	7 N Nitrogen 14.007 [He]2s <sup>2</sup> 2p <sup>3</sup>	8 O Oxygen 15.999 [He]2s <sup>2</sup> 2p <sup>4</sup>	9 F Fluorine 18.998 [He]2s <sup>2</sup> 2p <sup>5</sup>	10 Ne Neon 20.180 [He]2s <sup>2</sup> 2p <sup>6</sup>
2																		
3	11 Na Sodium 22.990 [Ne]3s <sup>1</sup>	12 Mg Magnesium 24.305 [Ne]3s <sup>2</sup>											13 Al Aluminum 26.982 [Ne]3s <sup>2</sup> 3p <sup>1</sup>	14 Si Silicon 28.086 [Ne]3s <sup>2</sup> 3p <sup>2</sup>	15 P Phosphorus 30.974 [Ne]3s <sup>2</sup> 3p <sup>3</sup>	16 S Sulfur 32.06 [Ne]3s <sup>2</sup> 3p <sup>4</sup>	17 Cl Chlorine 35.453 [Ne]3s <sup>2</sup> 3p <sup>5</sup>	18 Ar Argon 39.948 [Ne]3s <sup>2</sup> 3p <sup>6</sup>
4	19 K Potassium 39.098 [Ar]4s <sup>1</sup>	20 Ca Calcium 40.078 [Ar]4s <sup>2</sup>	21 Sc Scandium 44.956 [Ar]3d <sup>1</sup> 4s <sup>2</sup>	22 Ti Titanium 47.88 [Ar]3d <sup>2</sup> 4s <sup>2</sup>	23 V Vanadium 50.942 [Ar]3d <sup>3</sup> 4s <sup>2</sup>	24 Cr Chromium 51.996 [Ar]3d <sup>5</sup> 4s <sup>1</sup>	25 Mn Manganese 54.938 [Ar]3d <sup>5</sup> 4s <sup>2</sup>	26 Fe Iron 55.845 [Ar]3d <sup>6</sup> 4s <sup>2</sup>	27 Co Cobalt 58.933 [Ar]3d <sup>7</sup> 4s <sup>2</sup>	28 Ni Nickel 58.693 [Ar]3d <sup>8</sup> 4s <sup>2</sup>	29 Cu Copper 63.546 [Ar]3d <sup>10</sup> 4s <sup>1</sup>	30 Zn Zinc 65.38 [Ar]3d <sup>10</sup> 4s <sup>2</sup>	31 Ga Gallium 69.723 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup>	32 Ge Germanium 72.631 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup>	33 As Arsenic 74.922 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup>	34 Se Selenium 78.971 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup>	35 Br Bromine 79.904 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup>	36 Kr Krypton 84.98 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup>
5	37 Rb Rubidium 85.468 [Kr]5s <sup>1</sup>	38 Sr Strontium 87.62 [Kr]5s <sup>2</sup>	39 Y Yttrium 88.906 [Kr]4d <sup>1</sup> 5s <sup>2</sup>	40 Zr Zirconium 91.224 [Kr]4d <sup>2</sup> 5s <sup>2</sup>	41 Nb Niobium 92.906 [Kr]4d <sup>4</sup> 5s <sup>1</sup>	42 Mo Molybdenum 95.95 [Kr]4d <sup>5</sup> 5s <sup>1</sup>	43 Tc Technetium 98.907 [Kr]4d <sup>5</sup> 5s <sup>2</sup>	44 Ru Ruthenium 101.07 [Kr]4d <sup>7</sup> 5s <sup>1</sup>	45 Rh Rhodium 106.42 [Kr]4d <sup>8</sup> 5s <sup>1</sup>	46 Pd Palladium 106.42 [Kr]4d <sup>10</sup>	47 Ag Silver 107.868 [Kr]4d <sup>10</sup> 5s <sup>1</sup>	48 Cd Cadmium 112.414 [Kr]4d <sup>10</sup> 5s <sup>2</sup>	49 In Indium 114.818 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>1</sup>	50 Sn Tin 118.711 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup>	51 Sb Antimony 121.760 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup>	52 Te Tellurium 127.6 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup>	53 I Iodine 126.904 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup>	54 Xe Xenon 131.29 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup>
6	55 Cs Cesium 132.905 [Xe]6s <sup>1</sup>	56 Ba Barium 137.328 [Xe]6s <sup>2</sup>	57-71 Lanthanide Series	72 Hf Hafnium 178.49 [Xe]4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup>	73 Ta Tantalum 180.948 [Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup>	74 W Tungsten 183.84 [Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup>	75 Re Rhenium 186.207 [Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup>	76 Os Osmium 190.23 [Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup>	77 Ir Iridium 192.22 [Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup>	78 Pt Platinum 195.08 [Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>1</sup>	79 Au Gold 196.967 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>1</sup>	80 Hg Mercury 200.592 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup>	81 Tl Thallium 204.383 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>1</sup>	82 Pb Lead 207.2 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>2</sup>	83 Bi Bismuth 208.980 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>3</sup>	84 Po Polonium [209] [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>4</sup>	85 At Astatine [210] [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>5</sup>	86 Rn Radon 222.018 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup>
7	87 Fr Francium 223.020 [Rn]7s <sup>1</sup>	88 Ra Radium 226.025 [Rn]7s <sup>2</sup>	89-103 Actinide Series	104 Rf Rutherfordium 261 [Rn]5f <sup>14</sup> 6d <sup>2</sup> 7s <sup>2</sup>	105 Db Dubnium 262 [Rn]5f <sup>14</sup> 6d <sup>3</sup> 7s <sup>2</sup>	106 Sg Seaborgium 266 [Rn]5f <sup>14</sup> 6d <sup>4</sup> 7s <sup>2</sup>	107 Bh Bohrium 264 [Rn]5f <sup>14</sup> 6d <sup>5</sup> 7s <sup>2</sup>	108 Hs Hassium 277 [Rn]5f <sup>14</sup> 6d <sup>6</sup> 7s <sup>2</sup>	109 Mt Meitnerium 268 [Rn]5f <sup>14</sup> 6d <sup>7</sup> 7s <sup>2</sup>	110 Ds Darmstadtium 271 [Rn]5f <sup>14</sup> 6d <sup>8</sup> 7s <sup>2</sup>	111 Rg Roentgenium 272 [Rn]5f <sup>14</sup> 6d <sup>9</sup> 7s <sup>2</sup>	112 Cn Copernicium 285 [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup>	113 Uut Ununtrium [284] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>1</sup>	114 Fl Flerovium [289] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>2</sup>	115 Uup Ununpentium [288] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>3</sup>	116 Lv Livermorium [293] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>4</sup>	117 Uus Ununseptium [294] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>5</sup>	118 Uuo Ununoctium [294] [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>6</sup>

Lanthanide Series	57 La Lanthanum 138.905 [Xe]5d <sup>1</sup> 6s <sup>2</sup>	58 Ce Cerium 140.116 [Xe]4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	59 Pr Praseodymium 140.908 [Xe]4f <sup>3</sup> 6s <sup>2</sup>	60 Nd Neodymium 144.24 [Xe]4f <sup>4</sup> 6s <sup>2</sup>	61 Pm Promethium 144.913 [Xe]4f <sup>5</sup> 6s <sup>2</sup>	62 Sm Samarium 150.36 [Xe]4f <sup>6</sup> 6s <sup>2</sup>	63 Eu Europium 151.964 [Xe]4f <sup>7</sup> 6s <sup>2</sup>	64 Gd Gadolinium 157.25 [Xe]4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	65 Tb Terbium 158.925 [Xe]4f <sup>9</sup> 6s <sup>2</sup>	66 Dy Dysprosium 162.500 [Xe]4f <sup>10</sup> 6s <sup>2</sup>	67 Ho Holmium 164.930 [Xe]4f <sup>11</sup> 6s <sup>2</sup>	68 Er Erbium 167.259 [Xe]4f <sup>12</sup> 6s <sup>2</sup>	69 Tm Thulium 168.934 [Xe]4f <sup>13</sup> 6s <sup>2</sup>	70 Yb Ytterbium 173.055 [Xe]4f <sup>14</sup> 6s <sup>2</sup>	71 Lu Lutetium 174.967 [Xe]4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>
Actinide Series	89 Ac Actinium 227.028 [Rn]6d <sup>1</sup> 7s <sup>2</sup>	90 Th Thorium 232.038 [Rn]6d <sup>2</sup> 7s <sup>2</sup>	91 Pa Protactinium 231.036 [Rn]5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	92 U Uranium 238.029 [Rn]5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup>	93 Np Neptunium 237.048 [Rn]5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	94 Pu Plutonium 244.064 [Rn]5f <sup>6</sup> 7s <sup>2</sup>	95 Am Americium 243.061 [Rn]5f <sup>7</sup> 7s <sup>2</sup>	96 Cm Curium 247.070 [Rn]5f <sup>7</sup> 7s <sup>2</sup>	97 Bk Berkelium 247.070 [Rn]5f <sup>9</sup> 7s <sup>2</sup>	98 Cf Californium 251.080 [Rn]5f <sup>10</sup> 7s <sup>2</sup>	99 Es Einsteinium [252] [Rn]5f <sup>11</sup> 7s <sup>2</sup>	100 Fm Fermium 257.095 [Rn]5f <sup>12</sup> 7s <sup>2</sup>	101 Md Mendelevium 258.1 [Rn]5f <sup>13</sup> 7s <sup>2</sup>	102 No Nobelium 259.101 [Rn]5f <sup>14</sup> 7s <sup>2</sup>	103 Lr Lawrencium [262] [Rn]5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup>

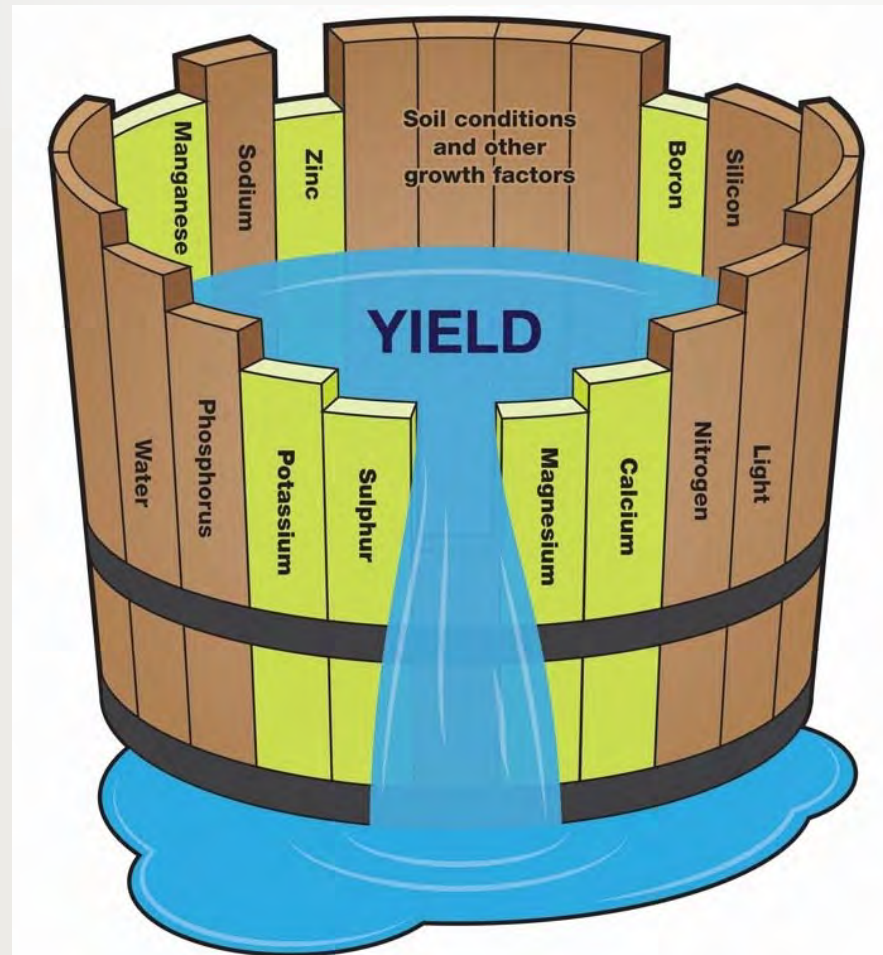
Alkali Metal	Alkaline Earth	Transition Metal	Basic Metal	Metalloid	Nonmetal	Halogen	Noble Gas	Lanthanide	Actinide
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# Plant-essential Nutrients

- “Macronutrients” are absorbed by plants in the largest amounts
  - C, H, O come from the atmosphere and water
  - N, P, K, Ca, Mg, S
- “Micronutrients” are absorbed in smaller amounts
  - Fe, Mn, Cu, Zn, Mo, Ni, B, Cl
  - A few other elements are required by some but not all plants (example: Si)
- Essential nutrients function in specific plant processes and can't be substituted

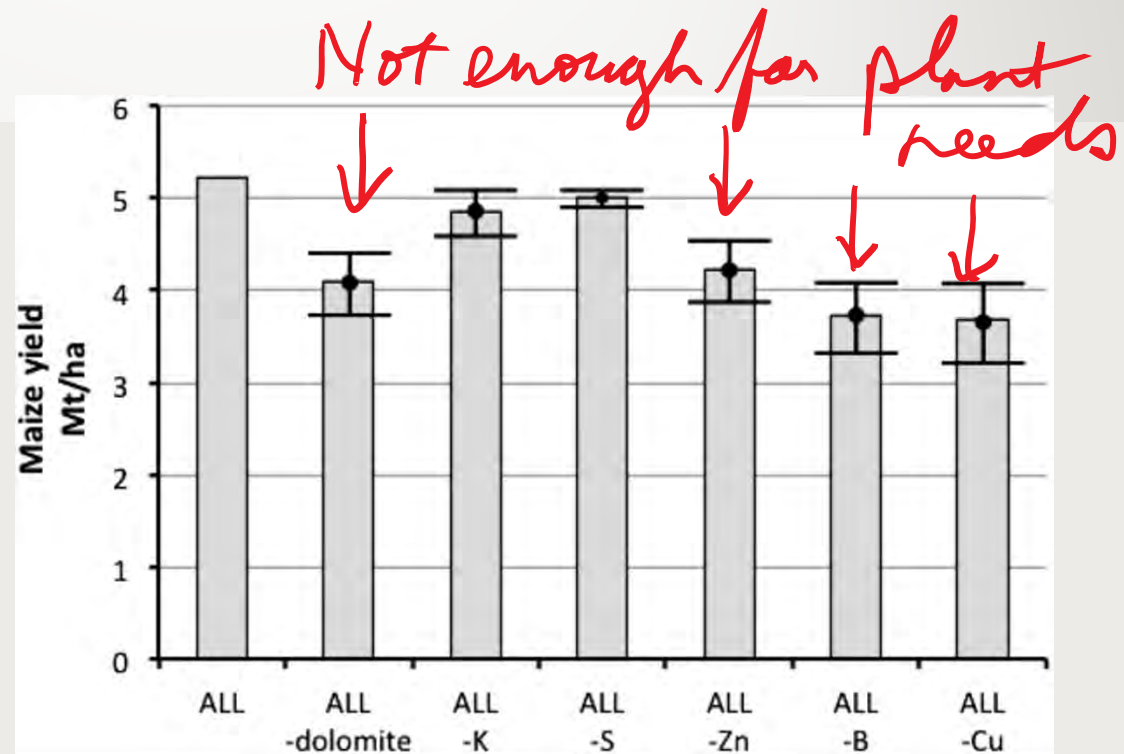
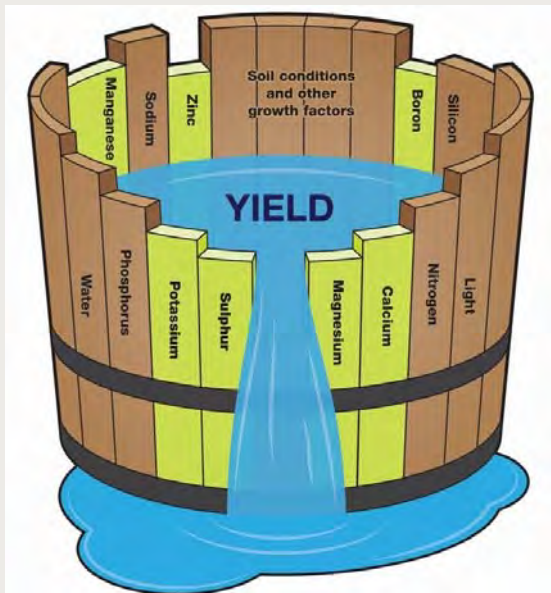


# Liebig's "law of the minimum"





# Liebig's "law of the minimum"



B. Vanlauwe et al., SOIL, 1, 491–508, 2015

[www.soil-journal.net/1/491/2015/](http://www.soil-journal.net/1/491/2015/)

Research in Burundi

# Nutrient Sources

- Weathering of naturally-occurring soil minerals
  - Most tropical soils are highly weathered and have low amounts of nutrient-bearing minerals (esp. P, K, Ca, Mg)
  - Nutrient release through soil mineral weathering is not a sustainable nutrient source!
- Nitrogen fixation from legume crops or cover crops
  - Biomass must stay on the soil for maximum N benefit
- Crop residues, manures, composts, etc.
- Fertilizers

# Plant available nutrients

- Nutrients contained in soil minerals or organic materials may not be available to plants. For some nutrients, only a tiny fraction is available.
- Plant available nutrients are nutrients soluble in water and accessible for uptake by plant roots.
- Plant available nutrients may be low because soils are nutrient-depleted. This is why we use fertilizers.
- Other factors—drought, soil pH, salinity may also limit nutrient availability

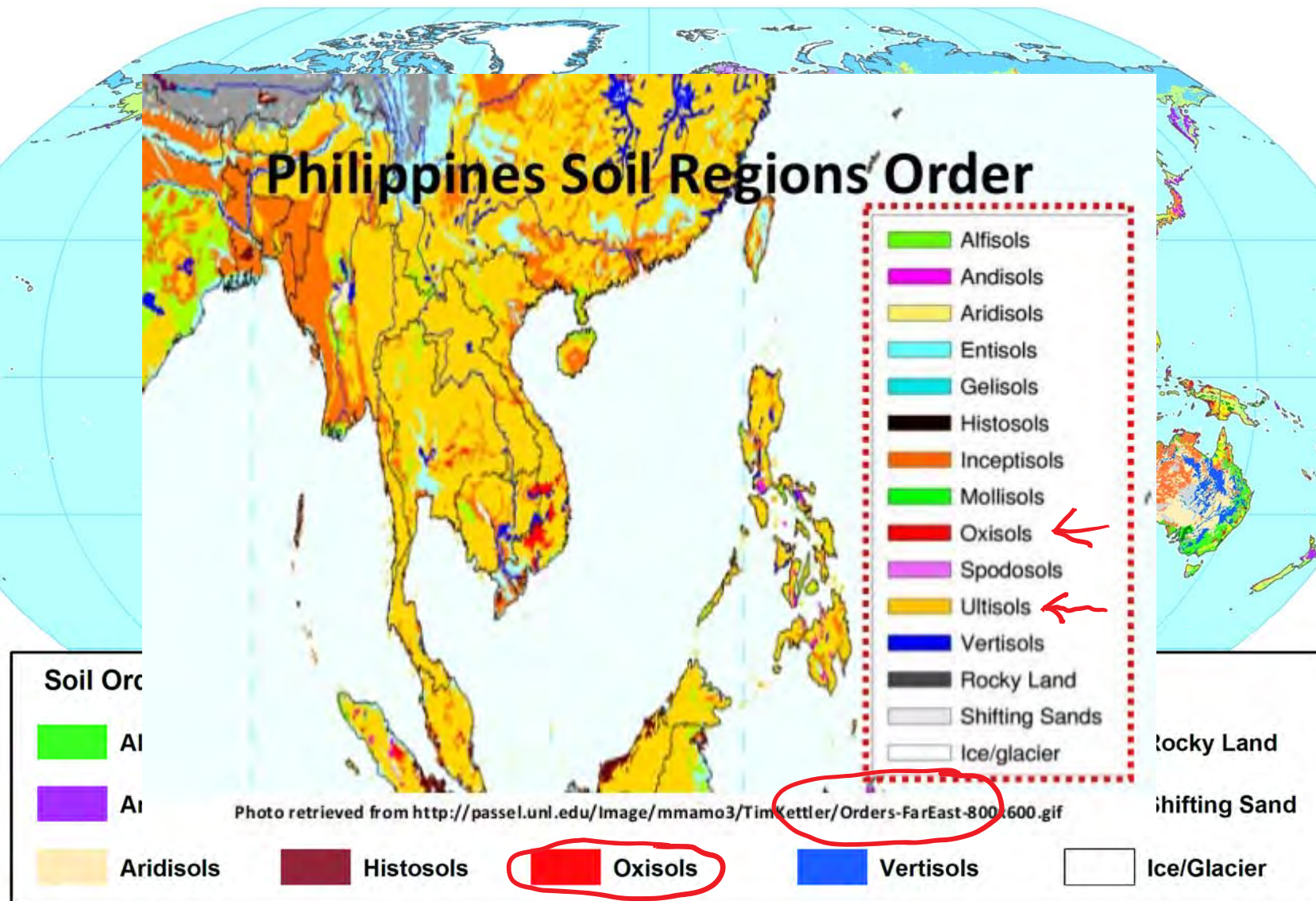


# The Five Soil-Forming Factors

- Parent Material (rocks)
- Climate
  - temperature, precipitation
- Organisms
  - Plants, microorganisms, animals, humans
- Topography (slope, aspect, elevation)
- Time



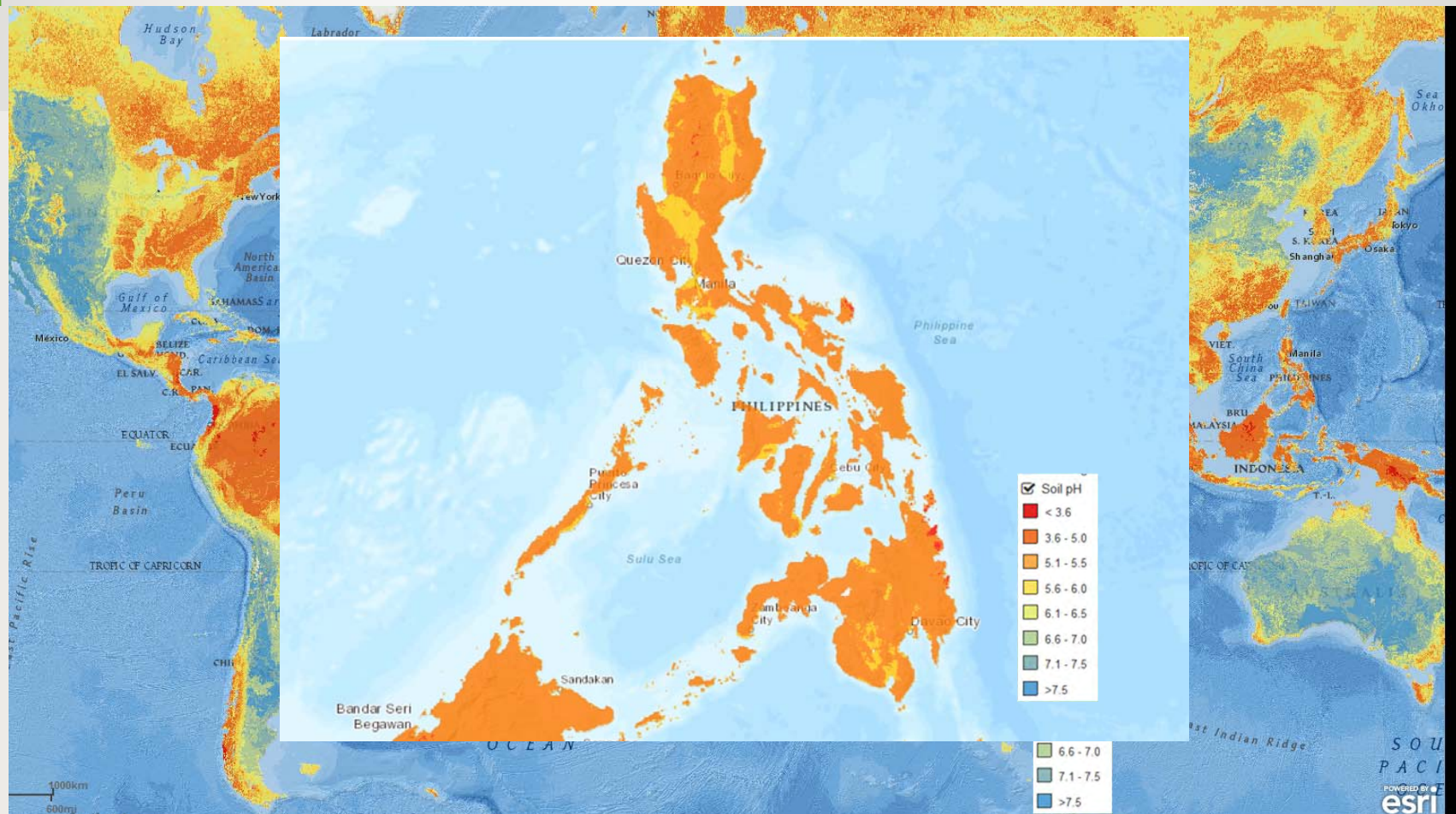




Ultisols commonly occur in moist subtropical climates. Highly leached, acid soils, often yellow-orange in color. Oxisols are most highly weathered soils, formed in humid tropical-subtropical climates. Acid, yellow-orange



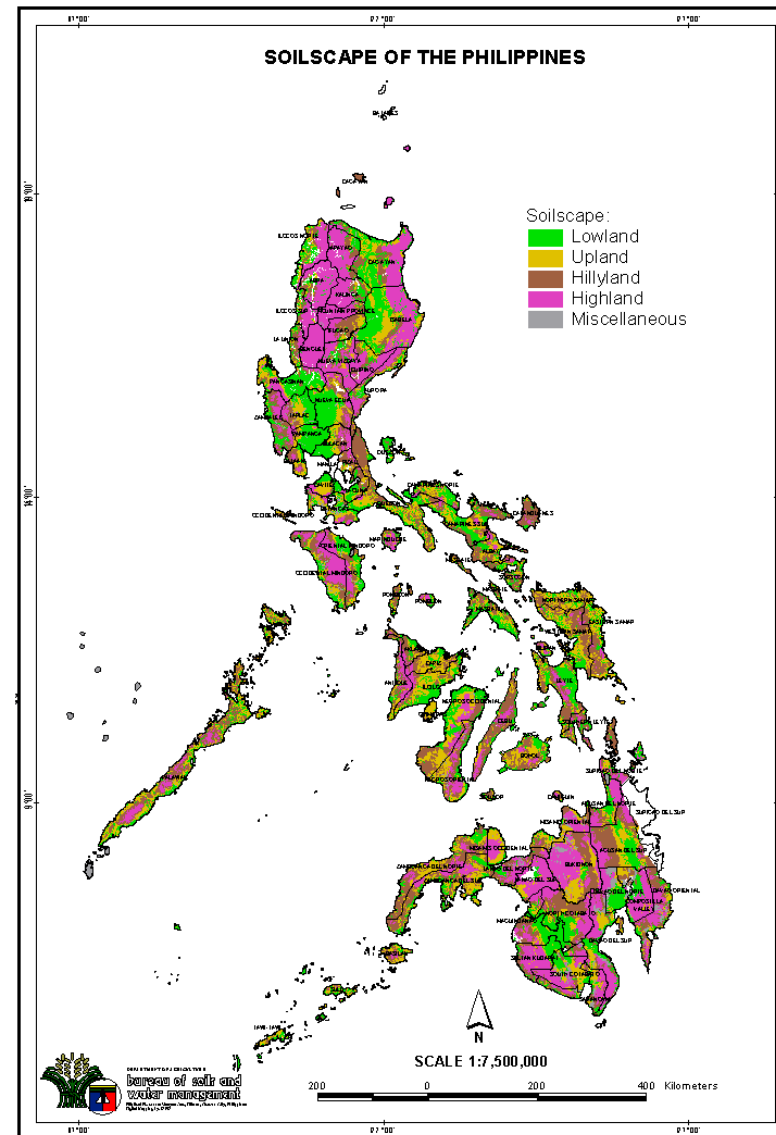
# pH of Tropical Soils



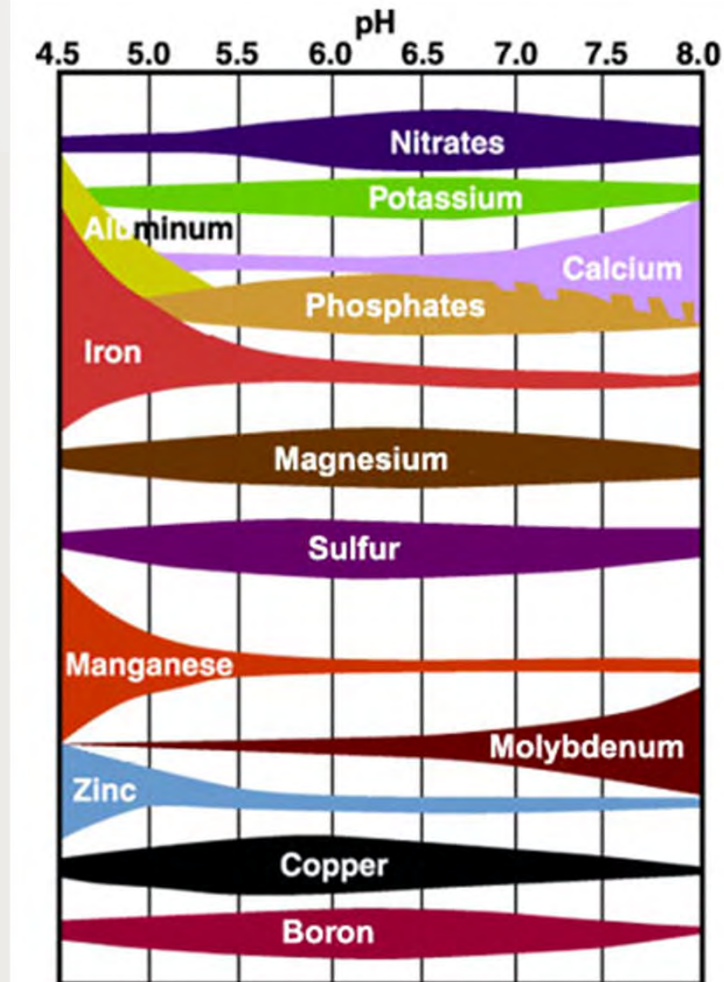
Global Agroclimatology Mapping, Virginia Tech

# Local Resources

- <http://bswm.da.gov.ph/>



# Soil pH and Nutrient Availability



<http://www.traylorchemical.com>



# Nitrogen

- The nutrient most often limiting to plant growth
- In soils, N exists mostly in soil organic matter and crop residues. This is one reason that it is so important to conserve and increase soil OM!
- N can be added to soils naturally through N fixation (bacterial process) with legumes used as crops, cover crops, and intercrops.

# Adding Nitrogen to Soils with Legumes

Cover crops



Intercropping with legumes



Rotating with legumes



# Phosphorus

- Phosphorus, among the most common elements found in the Earth's crust, was dubbed “life's bottleneck” by science writer Isaac Asimov. “[L]ife can multiply until all the phosphorus is gone, and then there is an inexorable halt which nothing can prevent. We may be able to substitute nuclear power for coal, and plastics for wood, and yeast for meat, and friendliness for isolation—but for phosphorus there is neither substitute nor replacement.”
- There is NO phosphorus equivalent to N fixation!



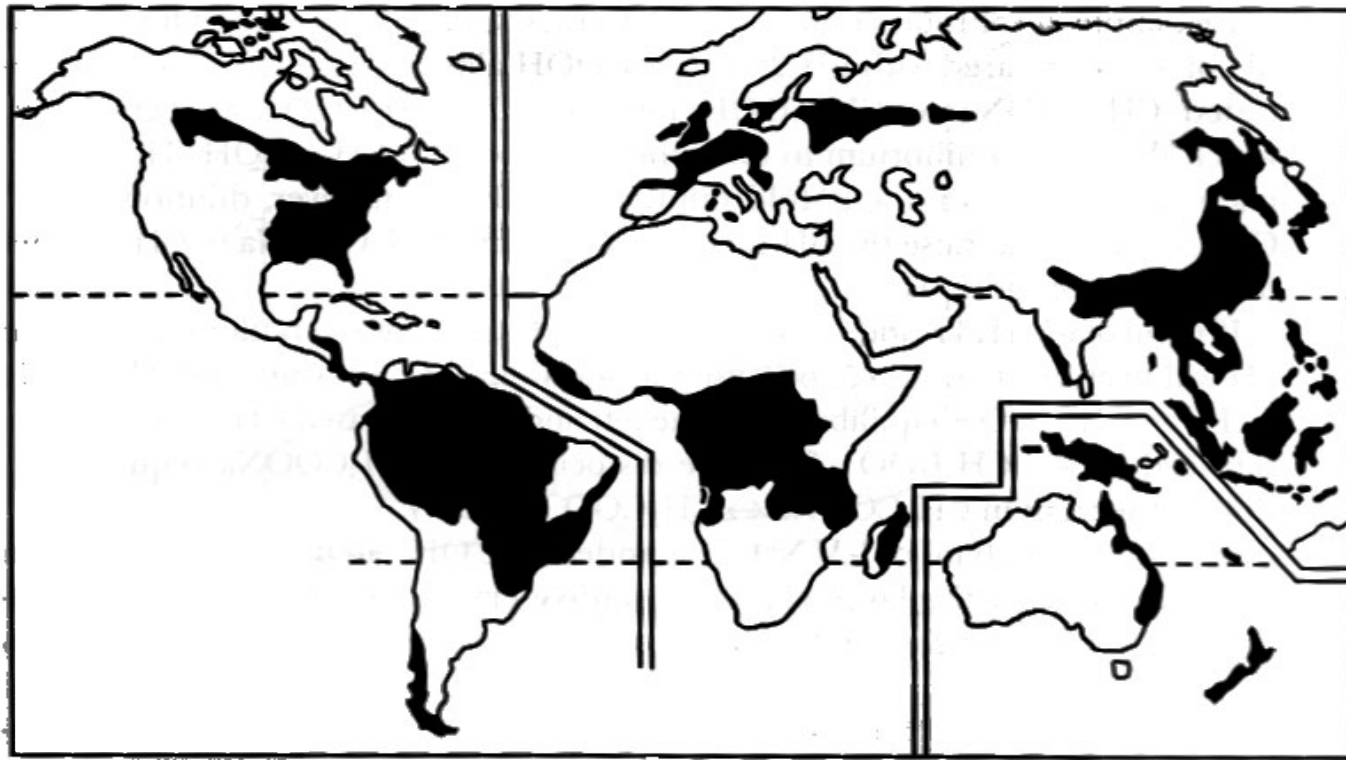
# Phosphorus deficiency



*Figure 7. Effect of bad  $P_2O_5$  supply (foreground) on crop development of forage maize.*

<https://livestocksystems.ilri.org/2016/01/27/manure-manual/>

## Soils low in K



## Common fertility properties of cropped tropical soils

- Acid pH of 5.5 or below and aluminum toxicity
- Low soil organic matter and N
- Low plant-available P (except where manures have been added)
- Low plant available K
- Low available B and Cl (except near coastlines)
- High availability of some micronutrients

# Nutrient Balance for Crop Production



## Nutrient Removal in Crop Harvest (kg/ha nutrient per t/ha grain yield)

Crop	N	P	K
Maize	12	3	4
Beans	50	6	13
Millet	28	4	7
Sorghum	13	3	5
Rice	13	3	3
Peanuts	35	3	7
Wheat	19	4	4
Banana (25 ton/ha)	45	6	110

## Nutrient Removal in Residue Harvest (kg/ha nutrient per t/ha grain yield)

Crop	N	P	K
Maize	8	1	17
Millet	14	2	17
Sorghum	13	3	5
Rice	8	1	17
Peanuts	16	2	10
Wheat	12	1	17

# Examples—nutrient removal

- Sub-Saharan Africa: Millet with 1.5 t/ha yield
  - 42 kg/ha N, 5 kg/ha P, 10 kg/ha K removed in grain
  - 21 kg/ha N, 3 kg/ha P, 25 kg/ha K removed in harvested residue
- SE Asia: Rice with 3.0 t/ha yield
  - 39 kg/ha N, 9 kg/ha P, 9 kg/ha K removed in grain
  - 24 kg/ha N, 3 kg/ha P, 51 kg/ha K removed in harvested (or burned) residue
- South America: Potato with 13 t/ha yield
  - 39 kg/ha N, 9 kg/ha P, 70 kg/ha K removed in tubers
  - 26 kg/ha N, 6 kg/ha P, 39 kg/ha K removed in residue
- Sub-Saharan Africa: Peanut with 1.0 t/ha yield
  - 35 kg/ha N, 3 kg/ha P, 7 kg/ha K removed in nuts/shells
  - 16 kg/ha N, 2 kg/ha P, 10 kg/ha K removed in harvested residue

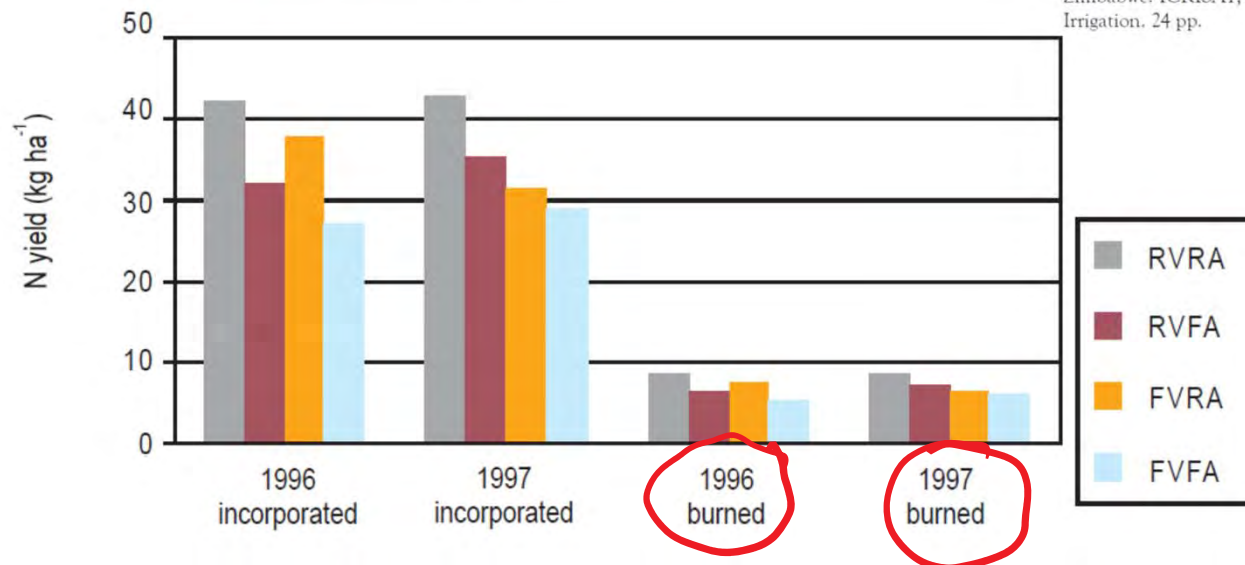
Source: IPNI Nutrient Calculator, <https://www.ipni.net/app/calculator/home>



# Residue conservation for nutrient benefit

Figure 5. Effect of variety, agronomy, and residue management on nitrogen contribution from groundnut residues.

Citation: ICRISAT/MAI. 2000. Cost-effective soil fertility management options for smallholder farmers in Malawi. PO Box 776, Bulawayo, Zimbabwe: ICRISAT; and Lilongwe, Malawi: Ministry of Agriculture and Irrigation. 24 pp.



RVRA = Researcher variety researcher agronomy, RVFA = Researcher variety farmer agronomy, FVRA = Farmer variety researcher agronomy etc.

Varieties: improved variety CG 7, local variety Chalimbana. Agronomic practices: number of weedings, plant population density.

The biggest factor in N benefit in both years was not variety or agronomy, but residue management, i.e. whether residues were burned or incorporated.

# Nutrient additions and losses

- Plant available nutrients are naturally lost from soils by...
  - Chemical weathering by rainfall
  - Erosion (soil loss)
  - Crop harvest and livestock grazing
- Plant available nutrients are naturally added to soils by...
  - Nitrogen fixation with legumes
  - Conservation of crop residues. [remember that these nutrients came from the soil to begin with—recycling rather than addition]
  - Manure or compost additions

# The nutrient “bottom line”

- Soils used for crop production experience net loss of plant available nutrients unless...
  - Legumes are part of a crop rotation (only effective for N)
  - Manures or other organic materials are IMPORTED from outside the field
  - Fertilizers are added
- Nutrient replacement is essential for sustainable crop production!
- Although soils differ in native fertility, there are NO soils that can support sustainable crop yields indefinitely without nutrient addition.



# Nutrient depletion most severe in Africa

- Annual nutrient depletion rates on cultivated lands during the past 30 years in 37 African countries:
  - 22 kg/ha of nitrogen (N),
  - 2.5 kg of phosphorus (P),
  - and 15 kg/ha of potassium (K)
- An annual loss equivalent to U.S. \$4 billion in fertilizer

Soil Fertility and Hunger in Africa, Pedro A. Sanchez (March 15, 2002), *Science* 295 (5562),

# Improving soil fertility in smallholder farming

- There is no magic!  
(Walang magic!)



# Building soil fertility in smallholder farming

- Avoiding nutrient losses
  - Reduce soil erosion with continuous soil cover
  - Return the highest possible amount of crop residue to soil
- Add nutrients to soil
  - Maximize N fixation with legumes in crop rotation, as cover crops and green manures (keep residues on soil!)
  - Intercropping with legume crops, trees, forbs
  - Return manures and ashes to soils
  - Fertilize using the right rate, time, place, and source (4R)
- Minimize other limitations to crop growth such as acidity, salinity, etc.
- “Conservation Agriculture”



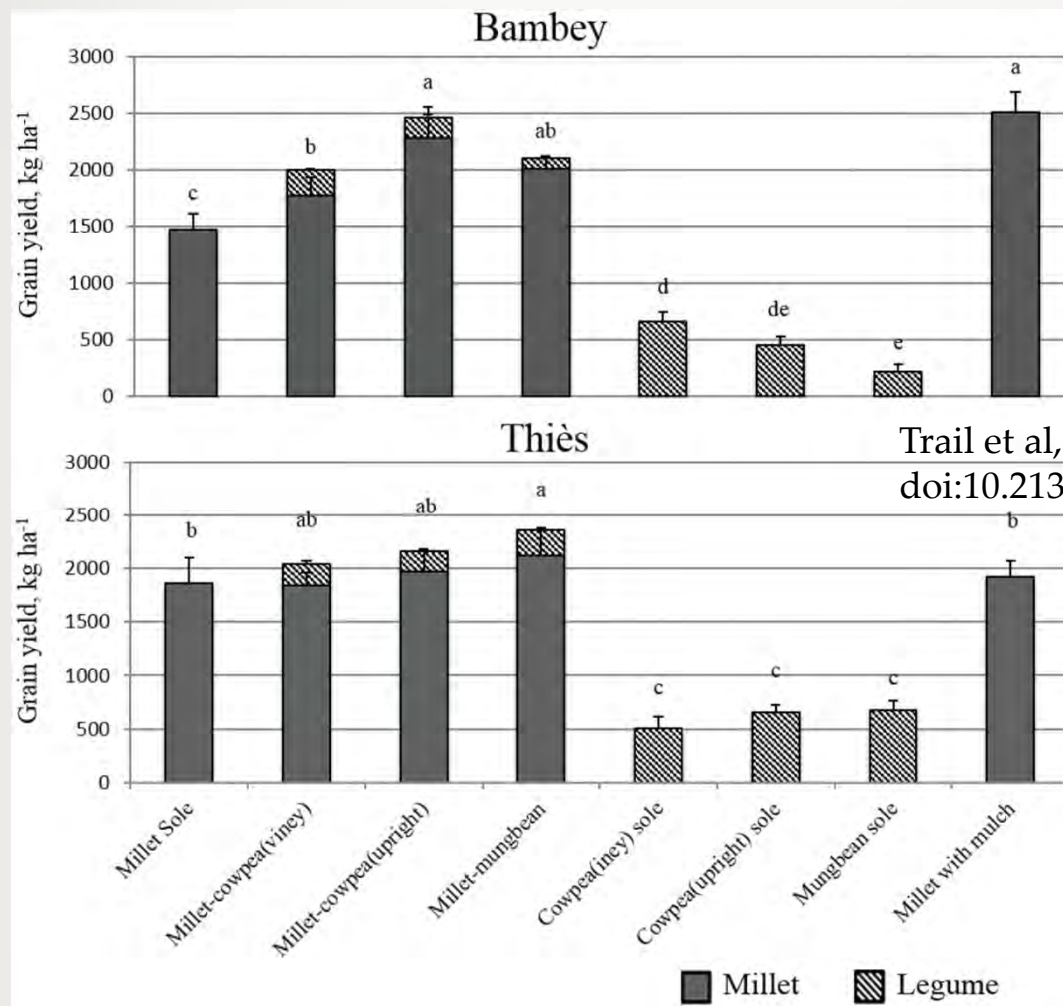
Protect the soil! Keep it covered!



# Building soil fertility in smallholder farming

- Avoiding nutrient losses
  - Reduce soil erosion with continuous soil cover
  - Return the highest possible amount of crop residue to soil
- Add nutrients to soil
  - Maximize N fixation with legumes in crop rotation, and as cover crops (keep residues on soil!)
  - Intercropping with legume crops, trees, forbs
  - Return manures and ashes to soils; add compost
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# Intercropping increased grain yields...



Trail et al, Agron. J. 108:1742–1752 (2016)  
doi:10.2134/agronj2015.0422



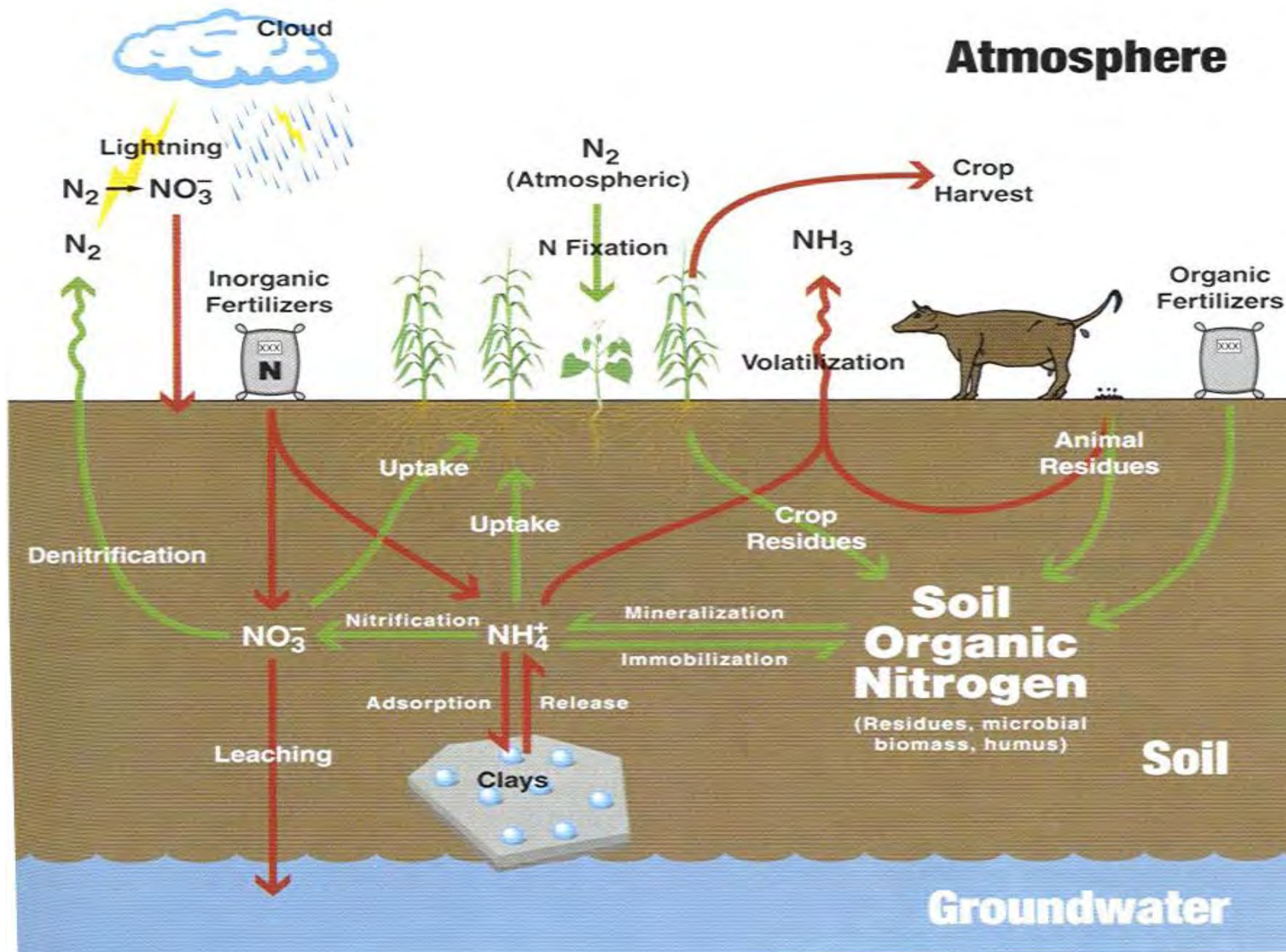
## The “four rights” approach for plant nutrients

- Apply at the right **AMOUNT** as established through soil and plant testing and recommendations, realistic yield goals, yield history, etc.
- Apply at the right **TIME** as established through knowledge of when plant nutrient demand is highest
- Apply at the right **PLACE** so that plant roots can access nutrients. Depends on crop, tillage, etc.
- Apply using the right **SOURCE** to maximize availability and minimize losses

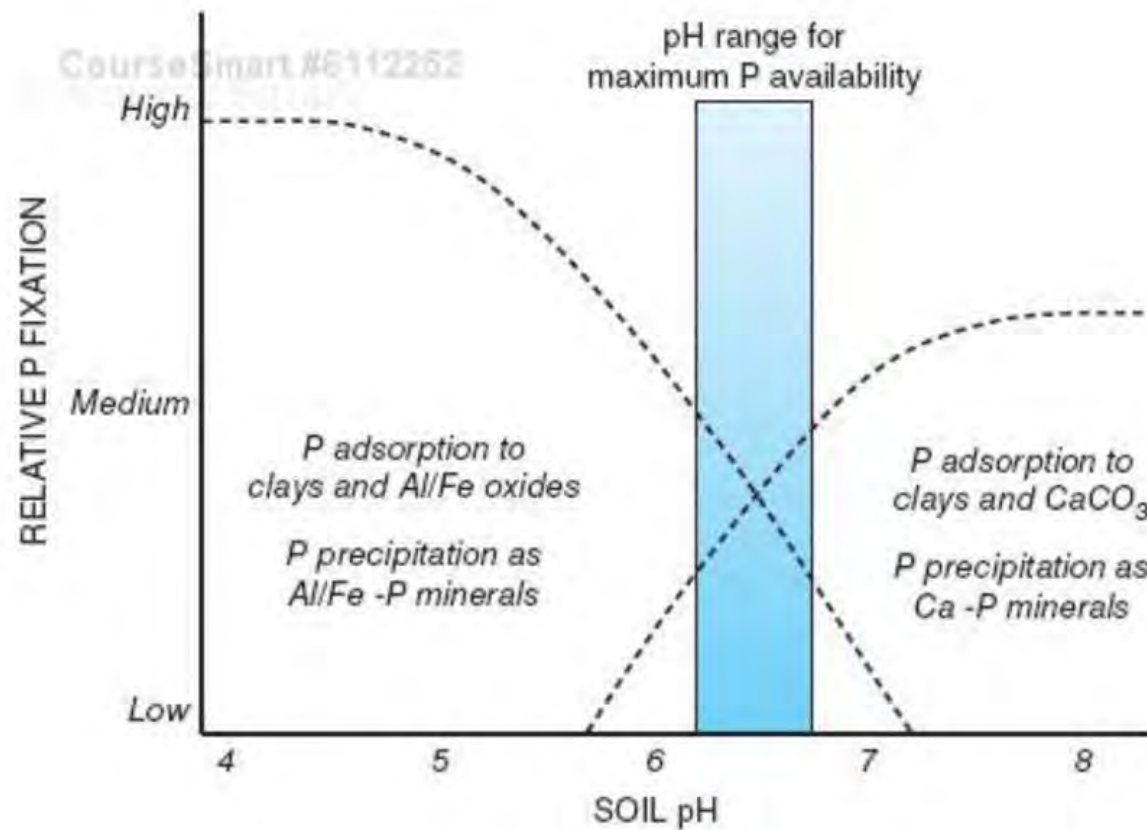
Good reference is 4R Plant Nutrient Management in African Agriculture,  
<http://ssa.ipni.net/ipniweb/region/africa.nsf>



# Atmosphere



# Soil pH and Phosphorus Availability



**Figure 5-11**  
Relative effect of soil pH on P adsorption and precipitation.

# P availability from fertilizers decreases with time

TABLE 5-11

CourseSmart #6112252

## INFLUENCE OF SOIL TYPE ON RESIDUAL FERTILIZER P AVAILABILITY

Soil Type	# of Soils	% P Available After 6 Months	
		Mean	Range
Calcareous	56	45	11-72
Slightly weathered	80	47	7-74
Moderately weathered	27	32	6-51
Highly weathered	40	27	14-54

Note: Resin extractable P measured 6 months after P application.

Source: Sharpley, 1991, Soil Sci. Soc. Am. J., 55:1038.



# What is a Fertilizer?

- A fertilizer is anything capable of providing plant-available nutrients
- Most fertilizers are synthetic (chemical) salts composed of plant nutrients
- Fertilizer sources:
  - N fertilizers from N in the atmosphere
  - P fertilizers are mined, treated to increase solubility
  - K fertilizers are mined and purified
- Organics can also be fertilizers



# Fertilizers

- Chemical or synthetic:
  - Not harmful to soils or soil organisms when used properly!
  - Supply nutrients directly in the forms that plants use
  - Although access and affordability can be a problem, chemical fertilizers are highly concentrated nutrient sources
- Organic:
  - Any organic material can be a fertilizer
  - Organic fertilizers provide soil organic matter also
  - Organic fertilizers contain mostly “slow-release” nutrients (N, P)
  - It is challenging to find enough organics to supply sustainable levels of nutrient addition for crop production, except on small scales

# Manure as N source



Figure 4. Available N from manure per ha grassland and arable crops and the N gain (dark topping) if losses are reduced from 40 to 30 %.

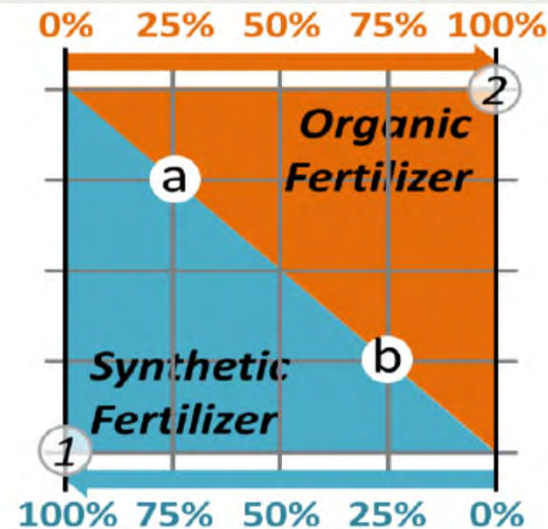


Figure 5. Organic fertilizer replacing synthetic fertilizers (read direction: bottom to up).

Excellent resource: <https://livestocksystems.ilri.org/2016/01/27/manure-manual/>

100

<http://sedac.ciesin.columbia.edu/data/collection/ferman-v1/maps/gallery/search>

# Composition of organic fertilizers

	% Moisture	N (% of dry weight)	P	K	N/P Ratio
Chicken	35	4.4	2.1	2.6	2.1
Cattle	80	1.9	0.7	2.0	2.7
Hog	72	2.1	0.8	1.2	2.6
Horse	63	1.4	0.4	1.0	3.5
Sheep	68	3.5	0.6	1.0	5.8
Municipal solid waste compost	40	1.2	0.3	0.4	4.0
Sewage sludge	80	4.5	2.0	0.3	2.25



# Nutrient release from manures

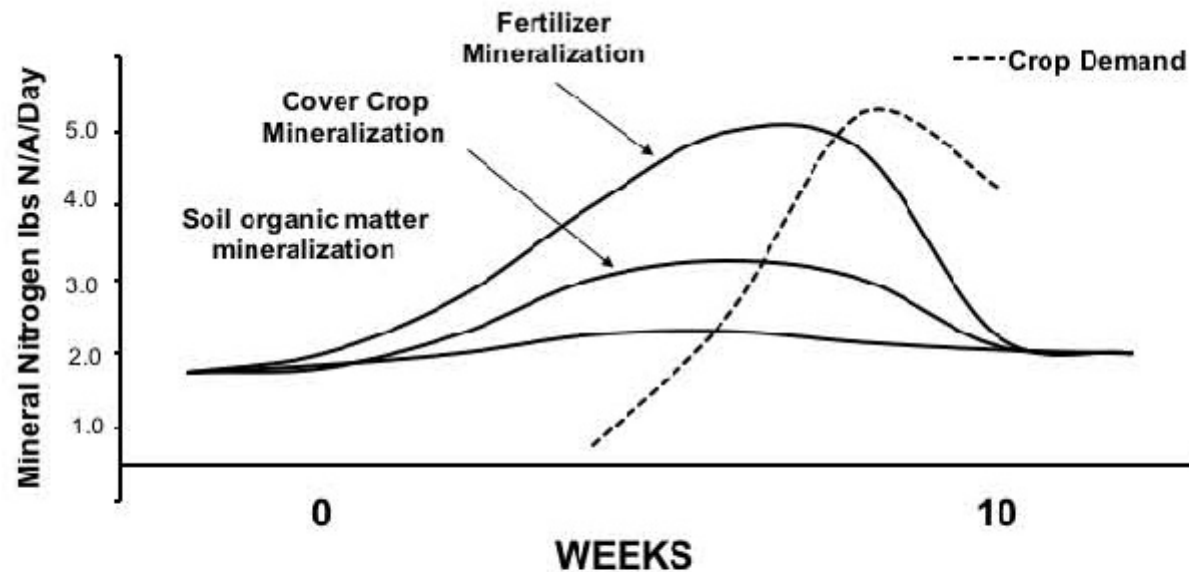
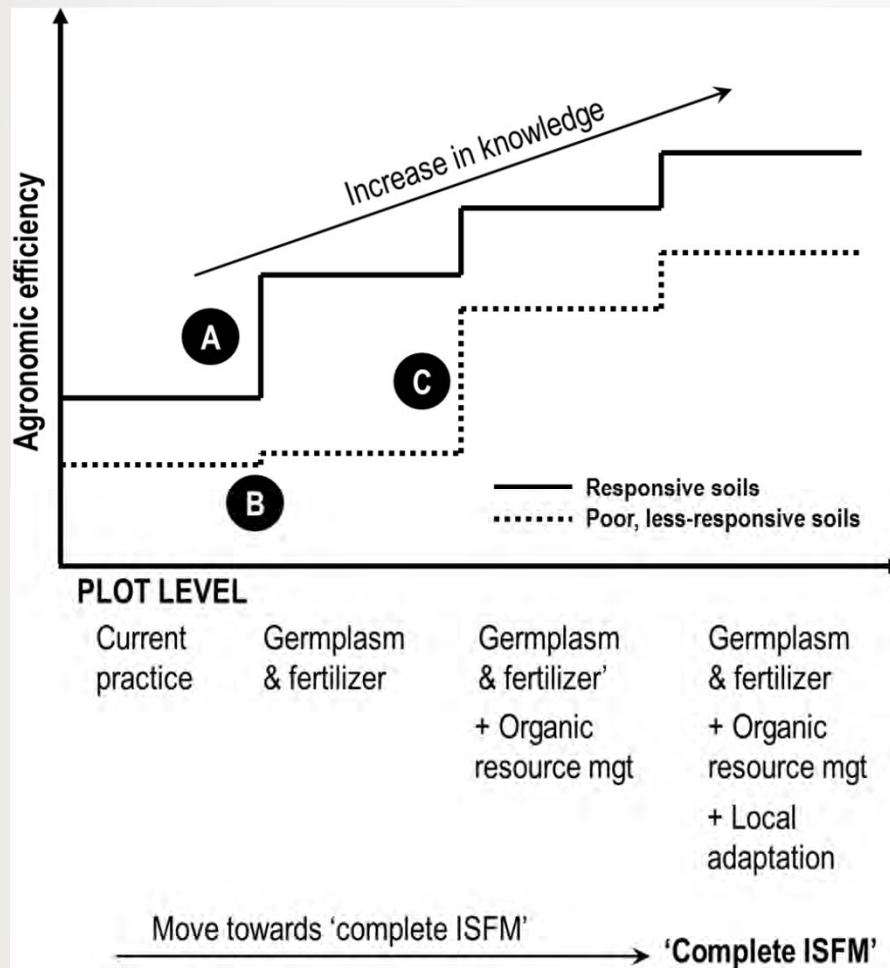


Figure 3. Theoretical depiction of N supplied to a crop from mineralization of soil organic matter, mineralization of crop residues and organic fertilizers

[http://ucanr.edu/blogs/salinasvalleyagriculture//blogfiles/43248\\_original.jpg](http://ucanr.edu/blogs/salinasvalleyagriculture//blogfiles/43248_original.jpg)

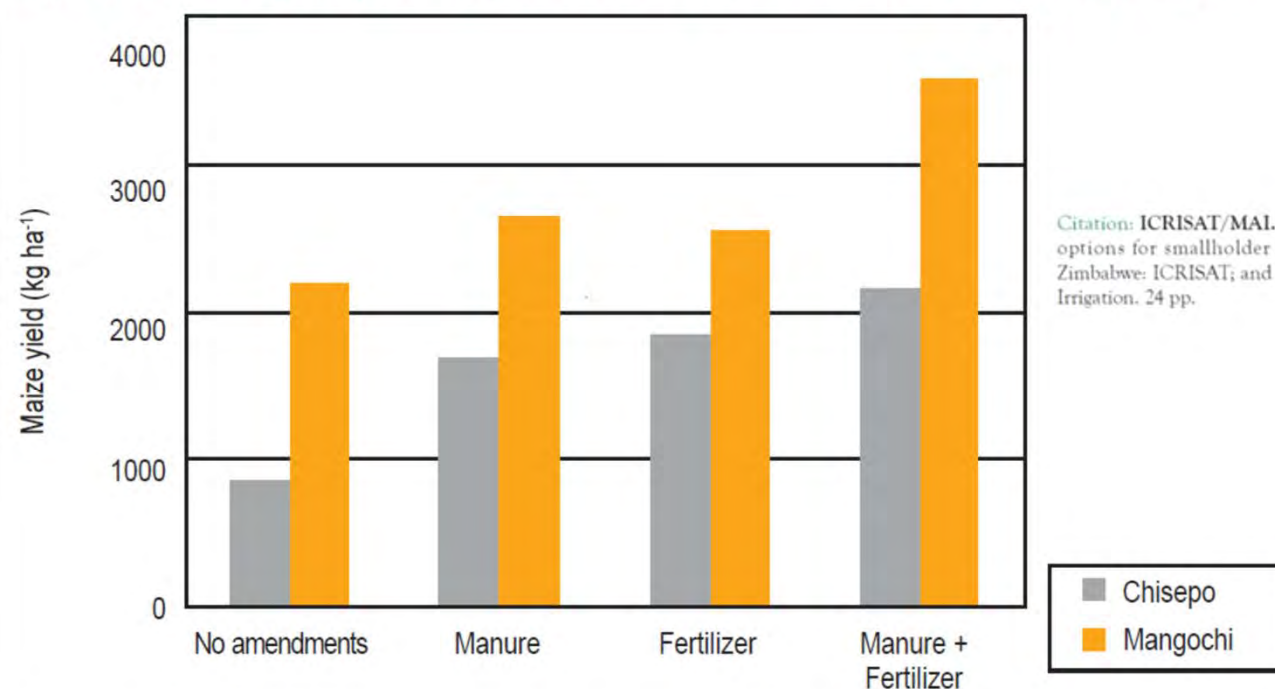
# Integrated Soil Fertility Management



B. Vanlauwe et al., SOIL, 1, 491–508, 2015  
[www.soil-journal.net/1/491/2015/](http://www.soil-journal.net/1/491/2015/)

# Integrated soil fertility mgmt.-example

Figure 3. Maize grain yield response to small amounts of manure and fertilizer.



Citation: ICRISAT/MAL. 2000. Cost-effective soil fertility management options for smallholder farmers in Malawi. PO Box 776, Bulawayo, Zimbabwe: ICRISAT; and Lilongwe, Malawi: Ministry of Agriculture and Irrigation. 24 pp.

Goat manure 5 t ha<sup>-1</sup>, targeted application of 2 handfuls per planting station. N fertilizer 17 kg ha<sup>-1</sup>. Data averaged from 10 on-farm trials at Chisepo (Central Malawi) and Mangochi (Southern Malawi), 1998.

## Closing points

- Small-medium farms produce 50-75% of the world's food
- All essential nutrients are important!
- Soils of the humid and subhumid tropics are usually nutrient-depleted and often acid
- Livestock grazing and crop production remove nutrients. They must be replaced for sustainability
- Active management is needed to avoid nutrient depletion. Phosphorus is particularly challenging
- Building soil fertility is hard work! But worth it!



| If you have questions...

- Attend my workshop today or tomorrow
- Email to [tlthompsonlbb@gmail.com](mailto:tlthompsonlbb@gmail.com)